

CHEMISTRY**9701/51**

Paper 5 Planning, Analysis and Evaluation

October/November 2017

MARK SCHEME

Maximum Mark: 30

Published

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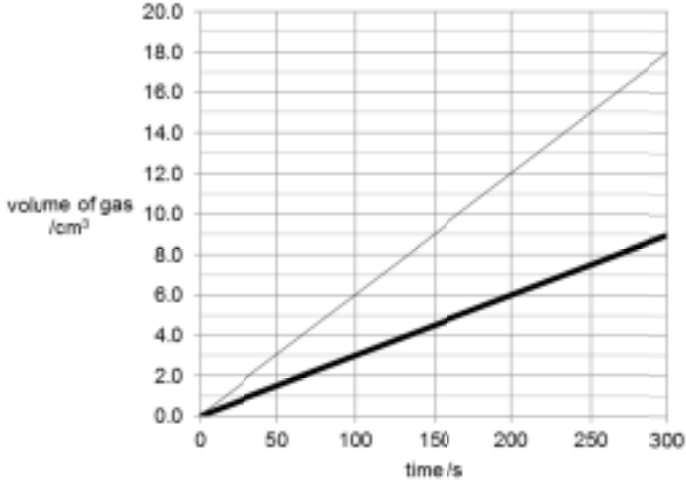
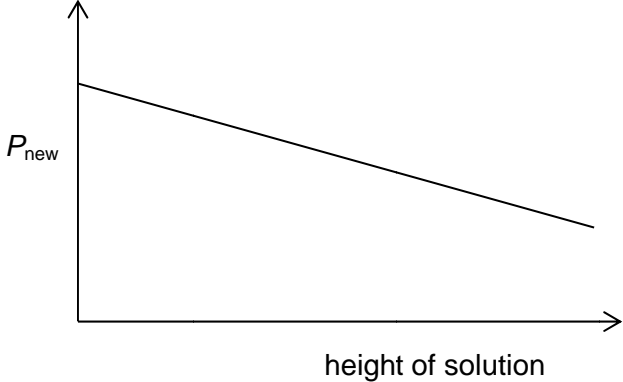
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This document consists of **6** printed pages.

Question	Answer	Marks
1(a)(i)	mass = $228.2 \times 1.00 \times (250/1000)$ = 57.1 g	1
1(a)(ii)	Distilled/deionised water must be mentioned somewhere for 2 marks to be given. Dissolve (all) the solid in a (suitable container) with (distilled) water	1
	Transfer / add to a <u>250 cm³</u> volumetric flask AND make to mark with (distilled) water	1
1(a)(iii)	(starch) gives a sharp 'end-point' / turns blue sharply / goes blue with volume of I ₂ invisible to naked eye	1
1(b)	volumes of (NH ₄) ₂ S ₂ O ₈ (aq) constant	1
	volumes of I ⁻ varying with range	1
	total volume constant, made up by water	1
1(c)(i)	mol I ₂ (aq) = $V(\text{S}_2\text{O}_3^{2-}) \times [\text{S}_2\text{O}_3^{2-}] / 2$ = $\left(\frac{0.005 \times 0.0050}{2} \right)$ = 1.25×10^{-5}	1
	$[\text{I}_2(\text{aq})] = \frac{\text{moles of I}_2(\text{aq})}{V_{\text{total}}} = \frac{1.25 \times 10^{-5}}{0.021} = 5.95 \times 10^{-4}$	1
	rate = $[\text{I}_2(\text{aq})] / \text{time}$ = $\frac{5.95 \times 10^{-4}}{134} = 4.44 \times 10^{-6}$	1
1(c)(ii)	repeat the experiment (and take average)	1
1(c)(iii)	% error = $\frac{2 \times 0.05}{5.0} \times 100\% = 2(.0) \%$	1

Question	Answer	Marks
1(d)	No thiosulfate had been added	1
1(e)	Ammonium persulfate must be stated along with its hazard and linked to the precaution. Ammonium persulfate is a skin irritant so wear gloves OR Ammonium persulfate is an irritant to the respiratory system; do the experiment in fume cupboard/face mask OR Ammonium persulfate is harmful if swallowed so avoid mouth contact/wear face mask OR Ammonium persulfate is oxidising so avoid contact with flammable/combustible materials. OR Ammonium persulfate is harmful/hazardous to the environment so do not dispose of down the drain/use (large quantities) of water to dilute before disposal	1

Question	Answer				Marks																																												
2(a)	<table border="1" data-bbox="342 220 1108 863"> <thead> <tr> <th data-bbox="342 220 533 355">time /s</th> <th data-bbox="533 220 723 355">burette reading /cm³</th> <th data-bbox="723 220 913 355">volume (of hydrogen) /cm³</th> <th data-bbox="913 220 1108 355">charge /C</th> </tr> </thead> <tbody> <tr><td>0</td><td>46.20</td><td>0.00</td><td>0</td></tr> <tr><td>50</td><td>41.20</td><td>5.00</td><td>40</td></tr> <tr><td>100</td><td>36.20</td><td>10.00</td><td>80</td></tr> <tr><td>150</td><td>31.45</td><td>14.75</td><td>120</td></tr> <tr><td>200</td><td>25.80</td><td>20.40</td><td>160</td></tr> <tr><td>250</td><td>20.80</td><td>25.40</td><td>200</td></tr> <tr><td>300</td><td>16.40</td><td>29.80</td><td>240</td></tr> <tr><td>350</td><td>11.45</td><td>34.75</td><td>280</td></tr> <tr><td>400</td><td>6.80</td><td>39.40</td><td>320</td></tr> <tr><td>450</td><td>1.50</td><td>44.70</td><td>360</td></tr> </tbody> </table> <p data-bbox="342 898 831 962">volumes of hydrogen correct to 2 d.p. charge correct</p>				time /s	burette reading /cm ³	volume (of hydrogen) /cm ³	charge /C	0	46.20	0.00	0	50	41.20	5.00	40	100	36.20	10.00	80	150	31.45	14.75	120	200	25.80	20.40	160	250	20.80	25.40	200	300	16.40	29.80	240	350	11.45	34.75	280	400	6.80	39.40	320	450	1.50	44.70	360	2
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2(b)	All ten points plotted correctly				1																																												
	Best-fit straight line drawn				1																																												
2(c)	Yes, (the data is reliable because) most of the points are on the line OR only a few points are not on the line.				1																																												
2(d)(i)	co-ordinates read and recorded correctly				1																																												
	gradient determined				1																																												
2(d)(ii)	= (i) ÷ 24000				1																																												
2(d)(iii)	= 1 ÷ (2 × (ii))				1																																												

Question	Answer	Marks
2(e)(i)	 <p>A line graph showing the volume of gas produced over time. The y-axis is labeled 'volume of gas /cm³' and ranges from 0.0 to 20.0 with major grid lines every 2.0 units and minor grid lines every 0.5 units. The x-axis is labeled 'time /s' and ranges from 0 to 300 with major grid lines every 50 units and minor grid lines every 10 units. Two straight lines originate from (0,0). The upper line passes through (300, 18.0). The lower line passes through (300, 9.0).</p> <p>straight line from origin to (300, 9.0)</p>	1
2(e)(ii)	Oxygen is (slightly) soluble in water	1
2(f)(i)	 <p>A graph showing the relationship between the new pressure P_{new} and the height of the solution. The y-axis is labeled P_{new} and the x-axis is labeled 'height of solution'. A straight line with a negative gradient is drawn, starting from a point on the y-axis and sloping downwards to the right.</p> <p>straight line with negative gradient</p>	1
2(f)(ii)	Faraday constant will be <u>lower</u> (than calculated) because the volume / V_m larger	1

Question	Answer	Marks
2(g)	No effect at cathode	1
	Less gas produced at anode	1
	Copper anode will dissolve/is (an) active (anode) OR copper has lower/more negative E^\ominus	1